

## Forskolin and 7-Deacetylforskolin: Study of the Reactivity Behaviour towards Different Phosphorylating Agents

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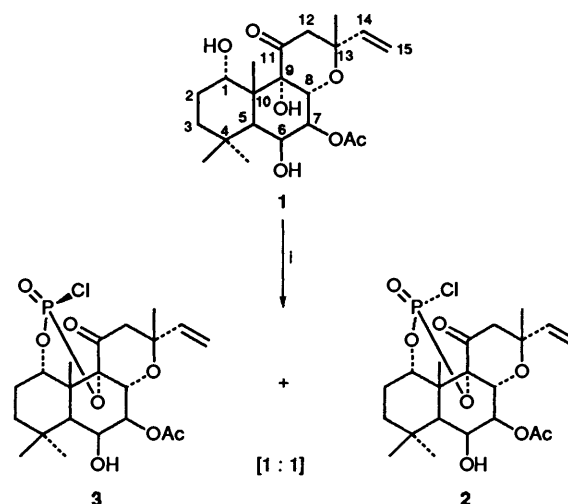
Reaction of different phosphorylating agents such as  $\text{POCl}_3$ , bis-(*p*-methoxyphenyl) phosphorochloridate, 2-chloro-2-oxo-1,3,2-dioxaphospholane and bis-(trifluoroethyl) phosphite with forskolin gave the corresponding isomers  $\alpha$ - and  $\beta$ -1, 9-cyclic phosphates/phosphites. 1,2-Dibromo-2-phenylethylphosphonic acid reacted with forskolin and 7-deacetylforskolin to give forskolin 1-phosphate **17** and 7-deacetylforskolin 7-phosphate **18**, respectively.

Forskolin, a labdane diterpenoid isolated from *Coleus forskohlii*,<sup>1</sup> shows a number of interesting biological properties, such as positive inotropic, antihypertensive, lowering of intraocular pressure and adenylate cyclase activator activities.<sup>2</sup> A large amount of chemical<sup>2a</sup> and biological<sup>2b</sup> data has been accumulated on forskolin and deoxyforskolin.

Forskolin is a polyhydroxylated labdane, and most of the hydroxy groups have been substituted with the aim of studying structure-activity relationships (SAR).<sup>3a</sup> The relative reactivity of different hydroxy groups based on their steric hindrance has, to a large extent, been delineated by different workers.<sup>3b</sup> The hydroxy group at C-7 for 7-deacetylforskolin is the most approachable; next available is the hydroxy group at C-1, followed by the highly hindered hydroxy group at C-6. We considered it a challenge and a useful exercise to try to phosphorylate specific hydroxy group/groups one at a time. Some of the resulting phosphates could be converted into water-soluble derivatives, a highly desirable feature for utilization of these compounds for different pharmacological studies. Forskolin by itself has very low water solubility and this has been one of the major drawbacks for the development of forskolin as a positive inotropic agent. Incidentally it may also be mentioned that only a limited amount of work on phosphorylation of terpenes in general has appeared. Consequently, we have synthesized phosphate and phosphite derivatives of forskolin and 7-deacetylforskolin at positions 1 and 7 as well as cyclic phosphates at positions 1, 9 and 6, 7, respectively. This paper describes this work.

### Results and Discussion

Forskolin **1** reacted with  $\text{POCl}_3$ ,  $\text{Et}_3\text{N}$  and pyridine in diethyl ether at 0–20 °C<sup>3a</sup> to give cyclic phosphorochloridates **2** and **3** (Scheme 1). The reaction is proposed to take place first at the OH group at position 1, and to be followed by cyclization through participation of the 9-OH group. It was observed that Cl groups on the new  $\text{OPOCl}_2$  intermediate showed no special discrimination for attack from 9-OH, therefore both  $\alpha$ -isomer **2** and  $\beta$ -isomer **3** were formed in a 1 : 1 mixture and were separated by flash chromatography on silica gel with  $\text{MeCN}-\text{C}_6\text{H}_6$  (15:85) as eluent. In the absence of phosphorus NMR, we have attempted to give a tentative stereochemical assignment of the  $\alpha$ -chloro **2** and  $\beta$ -chloro **3** based on their <sup>1</sup>H NMR data. In forskolin **1**, the 12 $\alpha$ -hydrogen is deshielded<sup>4</sup> and appears at low field compared with the 12 $\beta$ -hydrogen, due to the 1 $\alpha$ - and 9 $\alpha$ -OH groups. Substitution on these OH groups reduces the deshielding effect on the 12 $\alpha$ -proton, thereby narrowing the chemical-shift difference between the 12 $\alpha$  and 12 $\beta$  protons, as in compound **12a**<sup>5</sup> (Fig. 1). In cyclic  $\beta$ -phosphorochloridate **3**, the 12 $\beta$  and 12 $\alpha$  protons appeared at  $\delta$  2.74 and 3.16, respectively,



Scheme 1 Reagents: i,  $\text{POCl}_3$ ,  $\text{Et}_3\text{N}$ , pyridine,  $\text{Et}_2\text{O}$

leaving a difference in their chemical shifts of only 0.42 ppm. In the corresponding cyclic  $\alpha$ -phosphorochloridate **2** no decrease in the chemical-shift difference was observed. The 12 $\alpha$ -proton appeared at  $\delta$  3.50 and the 12 $\beta$ -H at  $\delta$  2.57, giving difference of 0.93 ppm in their chemical shifts. We have made an attempt to explain this phenomenon with the help of Dreiding models (Fig. 1) of the two isomers **2** and **3**. In isomer **2** the P=O bond is facing the 12 $\alpha$ -proton, which deshields this proton and mimics the deshielding effect of the 1,9-OH groups of forskolin **1**. However, in the isomer **3** the chloro group is  $\beta$ -oriented and the P=O bond occupies an  $\alpha$ -orientation. The P=O bond lies completely out of the plane of either the 12 $\alpha$ - or the 12 $\beta$ -hydrogen. Therefore no special deshielding effect is observed and the 12 $\alpha$ - and the 12 $\beta$ -proton appear close in their chemical shifts, as expected, because of cyclization (which we have observed when 1,9-OH groups are converted into cyclic structures such as acetonides)<sup>5</sup>. This phenomenon is common to all the pairs of isomers reported here and has been used as the basis for assigning the stereochemistry of different pairs of cyclic 1,9-phosphates of forskolin **1**.

Forskolin **1** reacted with bis-(*p*-methoxyphenyl) phosphorochloridate in pyridine<sup>6</sup> to give two cyclic phosphates,  $\alpha$ -isomer **4** and  $\beta$ -isomer **5**, in 56% yield, along with the minor amounts (1.6%) of the rearranged elimination product **6**. The yield of epoxide **6** could be increased to 13% upon replacement of pyridine with ethyldiisopropylamine (Hunig's base). The *p*-methoxyphenoxy group being a rather poor leaving group, it was expected that only the chloro group would be displaced to

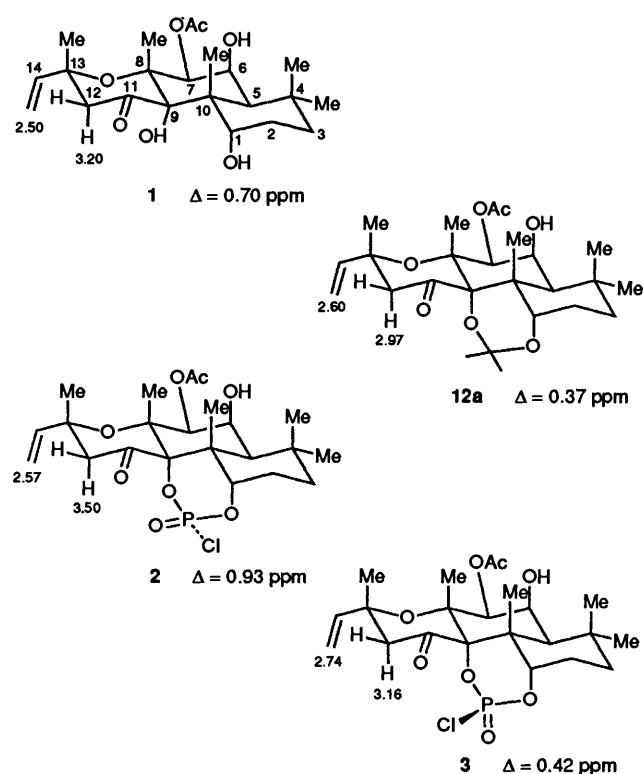


Fig. 1

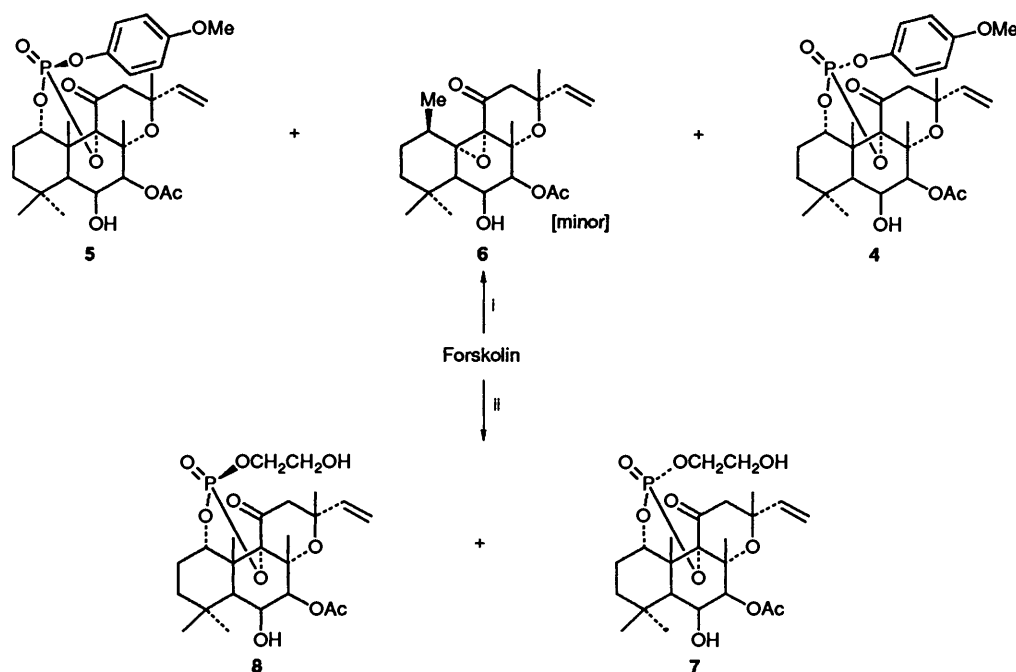
yield the phosphate at position 1. However, quite surprisingly, due to internal participation of the 9-OH group, a *p*-methoxyphenoxy group was displaced, giving rise to isomers 4 and 5 (Scheme 2). The structure of the rearranged elimination product 6 was assigned with the help of spectral data, the main characteristic  $^1\text{H}$  NMR peaks being: the 1-Me appeared at  $\delta$  0.74 as a doublet with  $J$  7.11 Hz, 1-H showed up as distorted sextet at  $\delta$  1.95, which became simplified when 1-Me was irradiated, and the 5-H proton gave a doublet of

6.1 Hz, indicating a change in the conformation of ring B. This ring presumably has assumed a distorted boat shape. The dihedral angle between 5-H and 6-H has considerably narrowed, thereby explaining the observed coupling constant.

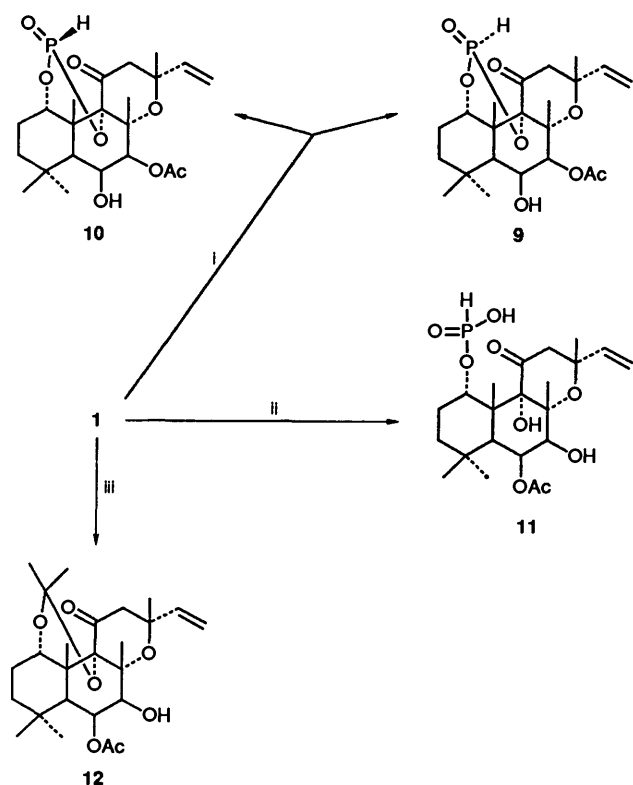
2-Chloro-2-oxo-1,3,2-dioxaphospholane<sup>7</sup> was allowed to react with forskolin 1 in pyridine in order to attempt reaction at position 1. However, this failed again due to internal participation of the 9-OH group and led to the formation of the corresponding cyclic phosphates,  $\alpha$ -isomer 7 and  $\beta$ -isomer 8 (Scheme 2). The assignment of the isomers is again based on the relative chemical shifts of the 12 $\beta$ - and 12 $\alpha$ -hydrogen (see Experimental section).

An interesting set of reactions took place when forskolin 1 reacted with bis(trifluoroethyl) hydrogen phosphite\*<sup>8</sup> under different conditions. In the presence of pyridine two isomeric cyclic phosphites (phosphonates),  $\alpha$ -isomer 9 and  $\beta$ -isomer 10, were isolated (Scheme 3). However, in benzene and without base the 1-mono(phosphonate) 11 was isolated as the sole product. The acetyl group at position 7 had migrated to C-6, as evidenced by its  $^1\text{H}$  NMR signal. Compound 11 was isolated as its diethylamine salt. When forskolin 1 reacted with bis(trifluoroethyl) hydrogen phosphite in acetone, again without base, no phosphorylation took place. Instead the acetonide 12 was isolated. We can offer some explanation for the formation of products 9–12. In the first reaction pyridine acts both as catalyst and as solvent. The phosphorylating reagent behaves in a manner similar to those discussed earlier to give the two cyclic isomers 9 and 10. Bis(trifluoroethyl) hydrogen phosphite reagent is acidic and in benzene condensation at C-1 is favoured. No participation by the 9-OH group is seen. Furthermore the anhydrous acidic environment can also favour the migration of the 7-acetyl group to the 6-position to give compound 11. Finally, with acetone as the solvent, the phosphorylating reagent acts as an acid catalyst and helps acetone to react with the 1- and 9-OH groups to give the 1,9-acetonide 12 at a much faster rate compared with phosphorylation. The above three

\* Drawn as the tautomeric form bis(trifluoroethyl) phosphonate in Scheme 3.



Scheme 2 Reagents: i, (*p*-MeOC<sub>6</sub>H<sub>4</sub>O)<sub>2</sub>P(=O)Cl, pyridine; ii, Cl(O=)POCH<sub>2</sub>CH<sub>2</sub>O, pyridine



**Scheme 3** Reagents: i,  $(\text{CF}_3\text{CH}_2\text{O})_2\text{P}(=\text{O})\text{H}$ , pyridine or  $\text{ClPOCH}_2\text{CH}_2\text{O}$ , pyridine; ii,  $(\text{CF}_3\text{CH}_2\text{O})\text{P}_2(=\text{O})\text{H}$ , benzene; then Zn, pyridine, water; iii,  $(\text{CF}_3\text{CH}_2\text{O})_2\text{P}(=\text{O})\text{H}$ , acetone

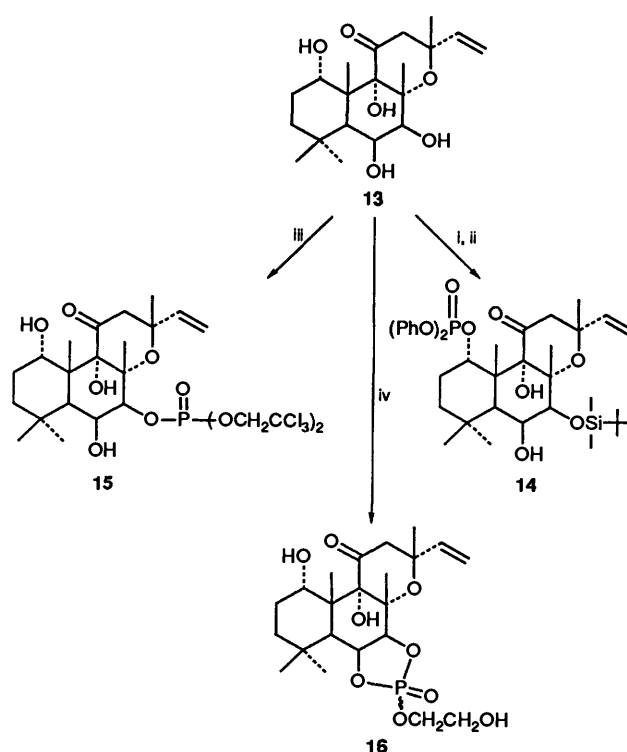
reactions clearly indicate that the phosphorylation of forskolin is sensitive to the overall reaction conditions.

Isomers **9** and **10** were treated separately with *N*-chlorosuccinimide (NCS) in  $\text{C}_6\text{H}_6$ , and the corresponding chloro compounds isolated were identical with compounds **2** and **3**, respectively. Formation of these chloro compounds proceeded with retention of their respective configuration.

Forskolin **1** was hydrolysed to 7-deacetylforskolin **13** and this was allowed to react with one mole equivalent of 2-chloro-2-oxo-1,3,2-dioxaphospholane in the presence of pyridine to give 6,7-phospholane **16**. The 6- and 7-OH groups behaved in a manner similar to the 1- and 9-OH groups in compound **1**. We believe that condensation takes place at the less hindered 7-position followed by participation of the neighbouring 6-OH group. The original phospholane ring then opens up and gives the new phospholane **16**. The stereochemistry of the hydroxyethoxy group could not be determined due to overlap of the peaks in the  $^1\text{H}$  NMR spectrum.

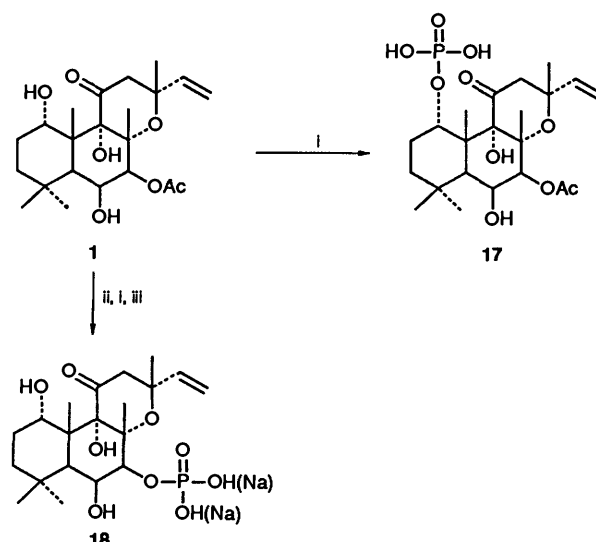
7-Deacetylforskolin **13**, on treatment with *tert*-butyldimethylsilyl chloride (TBDMSCl) and imidazole in dimethylformamide (DMF)<sup>9</sup> gave 7-(TBDMS)deacetylforskolin, which was then treated with diphenyl phosphorochloridate<sup>10</sup> in NaH-tetrahydrofuran (THF), to give only compound **14** (Scheme 4), and no cyclic product was detected. Reaction of compound **13** with bis-(2,2,2-trichloroethyl) phosphorochloridate<sup>11</sup> in pyridine gave compound **15**. All attempts to cleave the 2,2,2-trichloroethyl group gave a mixture of products.

1,2-Dibromo-2-phenylethylphosphonic acid is reported to generate 'PO<sub>3</sub><sup>-</sup>' species,<sup>12</sup> a controversial affair<sup>13</sup> which attracted our attention. We thought it would be worthwhile to attempt phosphorylation of forskolin **1** and 7-deacetylforskolin **13** with this reagent. Monophosphorylation at the 1- and 7-position, respectively, were achieved. Forskolin **1** was treated



**Scheme 4** Reagents: i, TBDMSCl, imidazole, DMF; ii,  $(\text{PhO})_2\text{P}(=\text{O})\text{Cl}$ , NaH; iii,  $(\text{CCl}_3\text{CH}_2\text{O})_2\text{P}(=\text{O})\text{Cl}$ , pyridine; iv,  $\text{Cl}(\text{O}=\text{P}(\text{O})\text{CH}_2\text{CH}_2\text{O}$ , pyridine

with 1,2-dibromo-2-phenylethylphosphonic acid in methylene dichloride and Hunig's base as catalyst. A relatively clean reaction took place to give the 1-phosphate **17** in 30% yield. Similarly, compound **13** reacted under identical conditions to give the corresponding 7-phosphate **18** in 29% yield (Scheme 5), which was further converted into its sodium salt by passage through Dowex 50X ( $\text{Na}^+$ ). It was observed that these reactions do not go to completion and almost 60% of the starting material



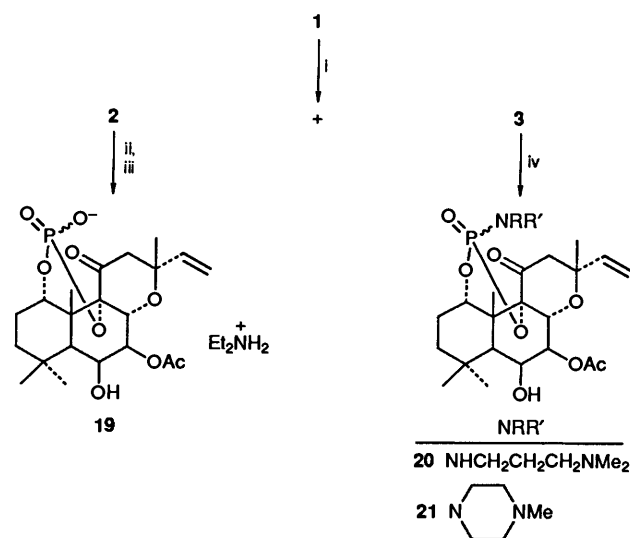
**Scheme 5** Reagents: i,  $\text{PhCHBrCHBrP}(=\text{O})(\text{OH})_2$ , Hunig's base; ii,  $^-\text{OH}$ ; iii, Dowex ( $\text{Na}^+$ )

was recovered. We have also noticed that the above reagent phosphorylates selectively. There is no need to protect any hydroxy groups, as was demonstrated with forskolin **1** and 7-deacetylforskolin **13**.

Attempted hydrolysis of compound **2** with water-1,4-dioxane

resulted in a mixture. This may be due to release of HCl, which leads to destruction of the molecule under reflux.<sup>14</sup> However, using Na<sub>2</sub>HPO<sub>4</sub> and KH<sub>2</sub>PO<sub>4</sub> as buffer along with water–1,4-dioxane produced hydrolysed product **19** in good yield. It was isolated as the diethylamine salt. The relative stereochemistry of the new OH group could not be determined by <sup>1</sup>H NMR spectroscopy because both OH and P=O groups can impart a deshielding effect on protons at C-12. More important is that presumably these are resonance structures. The chemical-shift difference between the 12 $\alpha$ - and 12 $\beta$ -proton is 1.39 ppm.

Condensation of  $\beta$ -chloro isomer **3** with 3-(dimethylamino)propylamine and with *N*-methylpiperazine in THF gave the corresponding phosphoramides **20** and **21**, respectively (Scheme 6).



**Scheme 6** Reagents: i, POCl<sub>3</sub>, Et<sub>3</sub>N, pyridine, Et<sub>2</sub>O; ii, H<sub>2</sub>O–dioxane, Na<sub>2</sub>HPO<sub>4</sub> and KH<sub>2</sub>PO<sub>4</sub>; iii, Et<sub>2</sub>NH; iv, HNRR'

In conclusion, phosphorylation follows the pattern of steric availability of the OH groups. In forskolin it is the 1-position which is phosphorylated first and generally through anchimeric assistance of 9-OH cyclic phosphates are formed; however, in deacetylforskolin **13** 7-OH is phosphorylated first. We have successfully synthesized the 1,9- and 6,7-cyclic phosphates and the 1- and 7-phosphates of forskolin and 7-deacetylforskolin. Some of the water-soluble compounds showed positive inotropic activity in guinea pig atria and others have intraocular-pressure-lowering properties and a few compounds have good blood-pressure-lowering properties. Details of biological activity will be reported elsewhere.

## Experimental

<sup>1</sup>H NMR spectra were recorded for solutions in CDCl<sub>3</sub> unless otherwise stated, with SiMe<sub>4</sub> internal standard, on Varian T-60 and JEOL-Fx-90 Q spectrometers. *J*-Values are given in Hz. IR spectra were recorded as KBr disks on Perkin-Elmer 157 and Perkin-Elmer 782 spectrometers. M.p.s were determined on a Kofler hot-stage melting point apparatus and are uncorrected. Light petroleum refers to the fraction boiling in the range 60–80 °C.

7- $\beta$ -Acetoxy-8,13-epoxy-6 $\beta$ -hydroxy-11-oxolabd-14-ene-1 $\alpha$ , 9 $\alpha$ -diyl Cyclophospho- $\alpha/\beta$ -chloridate **2** and **3**.—Forskolin **1** (6.15 g, 15 mmol) was dissolved in a mixture of diethyl ether (50 cm<sup>3</sup>), Et<sub>3</sub>N (2.1 cm<sup>3</sup>, 15 mmol) and pyridine (5 cm<sup>3</sup>) and the solution was chilled to –10 °C. A solution of POCl<sub>3</sub> (1.38 cm<sup>3</sup>, 15 mmol) in diethyl ether (10 cm<sup>3</sup>) was added dropwise for 15 min to the

chilled solution. The mixture was stirred for 1 h at –10 °C. The temperature was then allowed to rise to ambient and the mixture was stirred for 60 h. Solvent was removed under reduced pressure and the residue was taken up in EtOAc (300 cm<sup>3</sup>). The EtOAc layer was washed successively with cold, 1 mol dm<sup>-3</sup> HCl and water. It was then dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and the solvent was removed. The residue was purified by flash chromatography with 15% MeCN in C<sub>6</sub>H<sub>6</sub>, to give **compound 2** (2.45 g, 33%), m.p. 193–195 °C (EtOAc–light petroleum);  $\nu_{\max}/\text{cm}^{-1}$  3704, 3030, 1786, 1747, 1385 and 1136;  $\delta_{\text{H}}$  1.08 (3 H, s, Me), 1.3 (3 H, s, Me), 1.38 (3 H, s, Me), 1.61 (3 H, s, Me), 1.73 (3 H, s, Me), 1.88 (1 H, d, *J* 2.9, 5-H), 2.18 (3 H, s, Ac), 2.57 (1 H, d, *J*<sub>gem</sub> 18, 12 $\beta$ -H), 3.5 (1 H, d, *J*<sub>gem</sub> 18, 12 $\alpha$ -H), 4.48 (1 H, t, *J* 2.9, 6-H), 4.99 (1 H, dd, *J*<sub>cis</sub> 10.8, *J*<sub>gem</sub> 1.98, 15 H, Z), 5.20 (1 H, br, 1 H), 5.29 (1 H, d, *J* 3.6, 7-H), 5.29 (1 H, dd, *J*<sub>trans</sub> 18, *J*<sub>gem</sub> 1.98, 15 H, E) and 5.87 (1 H, dd, *J*<sub>trans</sub> 18, *J*<sub>cis</sub> 10.8, 14-H) (Found: C, 54.25; H, 6.5; Cl, 7.8; P, 6.1%. C<sub>22</sub>H<sub>32</sub>ClO<sub>8</sub>P requires C, 53.82; H, 6.57; Cl, 7.22; P, 6.31%), and **compound 3**, (0.9 g, 12%), m.p. 205 °C (decomp.) (from hot EtOAc),  $\nu_{\max}/\text{cm}^{-1}$  3509, 2941, 1730, 1718, 1299 and 1124;  $\delta_{\text{H}}$  1.09 (3 H, s, Me), 1.3 (3 H, s, Me), 1.34 (3 H, s, Me), 1.56 (3 H, s, Me), 1.78 (3 H, s, Me), 1.93 (1 H, d, *J* 2.9, 5-H), 2.15 (3 H, s, Ac), 2.74 (1 H, d, *J*<sub>gem</sub> 18.4, 12 $\beta$ -H), 3.16 (1 H, d, *J*<sub>gem</sub> 18.4, 12 $\alpha$ -H), 4.49 (1 H, t, *J* 2.9, 6-H), 5.08 (1 H, dd, *J*<sub>cis</sub> 10.8, *J*<sub>gem</sub> 1.86, 15-H Z), 5.19 (1 H, br, 1-H), 5.3 (1 H, dd, *J*<sub>trans</sub> 18, *J*<sub>gem</sub> 1.86, 15-H E), 5.32 (1 H, d, *J* 3.6, 7-H) and 5.89 (1 H, dd, *J*<sub>trans</sub> 18, *J*<sub>cis</sub> 10.8, 14-H) (Found: C, 53.8; H, 6.6; Cl, 7.6; P, 6.1%).

7 $\beta$ -Acetoxy-8,13-epoxy-6 $\beta$ -hydroxy-11-oxolabd-14-ene-1,9-diyl *p*-(Methoxyphenyl)-Phosphate **4** and **5**.—Forskolin **1** (3.28 g, 8 mmol) was dissolved in pyridine (30 cm<sup>3</sup>) and the solution was chilled to –10 °C. Bis-(*p*-methoxyphenyl)phosphorochloridate (4.37 g, 16 mmol) was added dropwise to the stirred solution. The reaction mixture was then gradually allowed to attain room temperature. The mixture was stirred overnight, taken up in EtOAc and washed successively with cold, 1 mol dm<sup>-3</sup> HCl and water. The EtOAc layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and the solvent was removed under reduced pressure. The residue was purified by flash chromatography with 50% EtOAc–light petroleum; **compound 6** was the first to elute, followed by  $\alpha$ -isomer **4** and  $\beta$ -isomer **5**.

**Compound 4** (1.28 g, 28%), m.p. 228 °C (from EtOAc–light petroleum);  $\nu_{\max}/\text{cm}^{-1}$  3419, 2941, 1717, 1493, 1356 and 1190;  $\delta_{\text{H}}$  0.51 (3 H, s, Me), 0.99 (1 H, d, *J* 2.7, 5-H), 1.16 (3 H, s, Me), 1.36 (3 H, s, Me), 1.5 (3 H, s, Me), 1.67 (3 H, s, Me), 2.25 (3 H, s, Ac), 2.56 (1 H, d, *J*<sub>gem</sub> 18, 12 $\beta$ -H), 3.6 (1 H, d, *J*<sub>gem</sub> 18, 12 $\alpha$ -H), 3.76 (3 H, s, OMe), 3.96 (1 H, t, *J* 3.6, 6-H), 4.96 (1 H, dd, *J*<sub>cis</sub> 10.8, *J*<sub>gem</sub> 1.8, 15-H, Z), 4.97 (1 H, br, 1-H), 5.21 (1 H, d, *J* 3.6, 7-H), 5.34 (1 H, dd, *J*<sub>trans</sub> 18, *J*<sub>gem</sub> 1.8, 15-H, E), 5.93 (1 H, dd, *J*<sub>trans</sub> 18, *J*<sub>cis</sub> 10.8, 14-H), 6.89 (2 H, d, *J* 9.9, ArH<sup>3,5</sup>) and 7.19 (2 H, d, *J* 9.9, ArH<sup>2,6</sup>) (Found: C, 60.1; H, 7.1; P, 5.4. C<sub>29</sub>H<sub>39</sub>O<sub>10</sub>P requires C, 60.20; H, 6.79; P, 5.35%).

**Compound 5** (1.31 g, 28%), m.p. 204 °C (EtOAc–light petroleum);  $\nu_{\max}/\text{cm}^{-1}$  3448, 2963, 1717, 1493, 1361, 1237 and 1194;  $\delta_{\text{H}}$  1.07 (3 H, s, Me), 1.3 (3 H, s, Me), 1.34 (3 H, s, Me), 1.57 (3 H, s, Me), 1.73 (3 H, s, Me), 1.87 (1 H, s, OH, exchanged with D<sub>2</sub>O), 2.06 (1 H, br, 5-H), 2.16 (3 H, s, Ac), 2.57 (1 H, d, *J*<sub>gem</sub> 18, 12 $\beta$ -H), 3.19 (1 H, d, *J*<sub>gem</sub> 18, 12 $\alpha$ -H), 3.76 (3 H, s, OMe), 4.47 (1 H, br t, *J* 3.15, 6-H), 4.91 (1 H, br, 1 H), 4.97 (1 H, dd, *J*<sub>cis</sub> 10.8, *J*<sub>gem</sub> 1.8, 15-H, Z), 5.27 (1 H, dd, *J*<sub>trans</sub> 18, *J*<sub>gem</sub> 1.8, 15-H, E), 5.41 (1 H, d, *J* 3.6, 7-H), 5.84 (1 H, dd, *J*<sub>trans</sub> 18, *J*<sub>cis</sub> 10.8, 14-H), 6.76 (2 H, d, *J* 9.9, ArH<sup>3,5</sup>) and 7.06 (2 H, d, *J* 9.9, ArH<sup>2,6</sup>) (Found: C, 60.0; H, 7.1; P, 5.2%).

**Compound 6** (0.05 g, 1.6%) with pyridine as the base and in 13% yield with Hunig's base, m.p. 158–164 °C;  $\nu_{\max}/\text{cm}^{-1}$  3448, 2923, 1706, 1315, 1257, 1068 and 985;  $\delta_{\text{H}}$  0.74 (3 H, d, *J* 7.11, 1-Me), 1.2 (6 H, s, 2  $\times$  Me), 1.35 (3 H, s, Me), 1.59 (3 H, s, Me), 2.13 (3 H, s, Me), 2.28 (3 H, s, Ac), 2.43 (1 H, d, *J*<sub>gem</sub> 13, 12 $\beta$ -H),

3.27 (1 H, d,  $J_{gem}$  13, 12 $\alpha$ -H), 4.24 (1 H, dd,  $J_{5,6}$  6.59,  $J_{6,7}$  4.6, 6-H), 4.97 (1 H, dd,  $J_{cis}$  10.56,  $J_{gem}$  1.62, 15-H, Z), 5.2 (1 H, dd,  $J_{trans}$  17.26,  $J_{gem}$  1.62, 15-H, E), 5.56 (1 H, d,  $J_{6,7}$  4.06, 7-H) and 5.87 (1 H, dd,  $J_{trans}$  17.26,  $J_{cis}$  10.56, 14-H) (Found: C, 67.8; H, 8.6. C<sub>22</sub>H<sub>32</sub>O<sub>6</sub> requires C, 67.32; H, 8.22%).

7 $\beta$ -Acetoxy-8,13-epoxy-6 $\beta$ -hydroxy-11-oxolabd-14-ene-1 $\alpha$ ,9 $\alpha$ -diyl 2-Hydroxyethyl- $\alpha/\beta$ -phosphate 7 and 8.—Forskolin 1 (1.23 g, 3 mmol) was dissolved in a mixture of CH<sub>2</sub>Cl<sub>2</sub> (20 cm<sup>3</sup>) and pyridine (2 cm<sup>3</sup>) and the solution was chilled to -10 °C. To this vigorously stirred solution was added 2-chloro-2-oxo-1,3,2-dioxaphospholane (2 cm<sup>3</sup>, 14 mmol) dropwise. After 1 h at 0 °C the reaction mixture was kept in the refrigerator (5 °C) for 24 h. It was then diluted with EtOAc and washed successively with cold, mol dm<sup>-3</sup> HCl and water. The EtOAc layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and the solvent was removed. The residue was purified by flash chromatography with 30% EtOAc–light petroleum ( $\alpha$ -isomer 7) followed by 20% MeCN–CHCl<sub>3</sub> ( $\beta$ -isomer 8).

**Compound 7** (0.31 g, 20%), m.p. 198–200 °C (from EtOAc–light petroleum);  $\nu_{max}/cm^{-1}$  3389, 2919, 1709, 1356, 1250, 1169 and 1090;  $\delta_H$  1.07 (3 H, s, Me), 1.30 (3 H, s, Me), 1.36 (3 H, s, Me), 1.59 (3 H, s, Me), 1.70 (3 H, s, Me), 1.93 (1 H, d,  $J$  2.7, 5-H), 2.17 (3 H, s, Ac), 2.54 (1 H, d,  $J_{gem}$  18, 12 $\beta$ -H), 3.51 (1 H, d,  $J_{gem}$  18, 12 $\alpha$ -H), 3.7 (2 H, t,  $J$  5.4, CH<sub>2</sub>CH<sub>2</sub>OH), 4.29 (2 H, dd,  $J$  10.8 and 5.4, OCH<sub>2</sub>CH<sub>2</sub>OH), 4.46 (1 H, t,  $J$  3.15, 6-H), 4.91 (1 H, dd,  $J_{cis}$  10.8,  $J_{gem}$  1.8, 15-H, Z), 5.01 (merged, 1 H, 1-H), 5.27 (1 H, dd,  $J_{trans}$  18,  $J_{gem}$  1.8, 15-H, E), 5.34 (1 H, br, 7-H) and 5.86 (1 H, dd,  $J_{trans}$  18,  $J_{cis}$  10.8, 14-H) (Found: C, 54.3; H, 7.4; P, 5.3. C<sub>24</sub>H<sub>37</sub>O<sub>10</sub>P·H<sub>2</sub>O requires C, 53.93; H, 7.35; P, 5.60%).

**Compound 8** (0.405 g, 26%), m.p. 182 °C (from EtOAc–light petroleum);  $\nu_{max}/cm^{-1}$  3401, 2941, 1709, 1351, 1285, 1273, 1235, 1116 and 1057;  $\delta_H$  1.09 (3 H, s, Me), 1.3 (3 H, s, Me), 1.36 (3 H, s, Me), 1.55 (3 H, s, Me), 1.7 (3 H, s, Me), 1.97 (1 H, d,  $J$  2.7, 5-H), 2.16 (3 H, s, Ac), 2.6 (1 H, d,  $J_{gem}$  18, 12 $\beta$ -H), 3.2 (1 H, d,  $J_{gem}$  18, 12 $\alpha$ -H), 3.61 (2 H, t,  $J$  5.4, CH<sub>2</sub>CH<sub>2</sub>OH), 4.2 (2 H, m, CH<sub>2</sub>CH<sub>2</sub>OH), 4.47 (1 H, t,  $J$  3.6, 6-H), 4.94 (1 H, br, 1-H), 4.97 (1 H, dd,  $J_{cis}$  10.8,  $J_{gem}$  1.8, 15-H, Z), 5.29 (1 H, dd,  $J_{trans}$  18,  $J_{gem}$  1.8, 15-H, E) 5.34 (1 H, br, 7-H) and 5.89 (1-H, dd,  $J_{trans}$  18,  $J_{cis}$  10.8, 14-H) (Found: C, 55.4; H, 7.3; P, 5.9%).

7 $\beta$ -Acetoxy-8,13-epoxy-6 $\beta$ -hydroxy-11-oxolabd-14-ene-1 $\alpha$ ,9 $\alpha$ -diyl Cyclo- $\alpha/\beta$ -phosphite 9 and 10.—Forskolin 1 (0.82 g, 2 mmol) was dissolved in dry pyridine (10 cm<sup>3</sup>) and to the vigorously stirred solution was added bis(trifluoroethyl) hydrogen phosphite (0.91 g, 6 mmol) under nitrogen. The mixture was stirred for 16 h at room temperature, taken up in EtOAc and washed successively with cold, 1 mol aq. dm<sup>-3</sup> HCl and water. The ethyl acetate layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and the solvent was removed. The solid residue was purified by flash chromatography with 30% EtOAc–light petroleum. Both  $\alpha$ -isomer 9 and  $\beta$ -isomer 10 were obtained pure.

**Compound 9** (0.36 g, 39%), m.p. 225–227 °C (from EtOAc–light petroleum);  $\nu_{max}/cm^{-1}$  3419, 2899, 1730, 1709, 1351, 1253, 1230, 1166, 1114 and 1087;  $\delta_H$  1.06 (3 H, s, Me), 1.36 (3 H, s, Me), 1.61 (6 H, s, 2 × Me), 1.71 (3 H, s, Me), 1.94 (1 H, OH, exchangeable), 2.17 (3 H, s, Ac), 2.52 (1 H, d,  $J_{gem}$  18, 12 $\beta$ -H), 3.62 (1 H, d,  $J_{gem}$  18, 12 $\alpha$ -H), 4.49 (1 H, t,  $J$  2.9, 6-H), 4.97 (1 H, dd,  $J_{cis}$  10.8,  $J_{gem}$  1.8, 15-H, Z), 5.24 (1 H, dd,  $J_{trans}$  18,  $J_{gem}$  1.8, 15-H), 5.11 and 5.34 (2 H, br hump, 1- and 7-H) and 5.91 (1 H, dd,  $J_{trans}$  18,  $J_{cis}$  10.8, 14-H) (Found: C, 58.1; H, 7.8; P, 6.4. C<sub>22</sub>H<sub>33</sub>O<sub>8</sub>P requires C, 57.89; H, 7.29; P, 6.79%).

**Compound 10** (0.12 g, 13%), m.p. 212 °C (from EtOAc–light petroleum);  $\nu_{max}/cm^{-1}$  3333, 2877, 1702, 1356, 1262, 1236, 1169 and 1090;  $\delta_H$  1.1 (3 H, s, Me), 1.30 (3 H, s, Me), 1.36 (3 H, s, Me), 1.50 (3 H, s, Me), 1.79 (3 H, s, Me), 1.96 (1 H, s, exchangeable, OH), 2.11 (1 H, d,  $J$  3.2, 5-H), 2.19 (3 H, s, Ac), 2.77 (1 H, d,  $J_{gem}$

18, 12 $\beta$ -H), 3.01 (1 H, d,  $J_{gem}$  18, 12 $\alpha$ -H), 4.51 (1 H, t,  $J$  3.2, 6-H), 5.03 (1 H, dd,  $J_{cis}$  10.8,  $J_{gem}$  1.8, 15-H, Z), 5.19 (1 H, br, 1-H), 5.29 (1 H, dd,  $J_{trans}$  18,  $J_{gem}$  1.8, 15-H, E), 5.38 (1 H, d,  $J$  3.6, 7-H) and 5.93 (1 H, dd,  $J_{trans}$  18,  $J_{cis}$  10.8, 14 H) (Found: C, 58.2; H, 7.75; P, 6.5%).

6 $\beta$ -Acetoxy-8-13-epoxy-7 $\beta$ ,9 $\alpha$ -dihydroxy-11-oxolabd-14-en-1 $\alpha$ -yl Dihydrogen Phosphite Diethylamine Salt 11.—Forskolin 1 (0.82 g, 2 mmol) was dissolved in dry benzene (10 cm<sup>3</sup>) and the solution was chilled to 0 °C, and bis(trifluoroethyl) hydrogen phosphite (1.0 cm<sup>3</sup>) was added dropwise to the vigorously stirred mixture at 0 °C. After 2 h at 0 °C the reaction mixture was brought to room temperature and kept for 18 h. The solvent was removed under reduced pressure and the residue was purified by flash chromatography with 25% EtOAc–light petroleum, resulting in the intermediate phosphite, (0.44 g), which is very unstable and immediately reacted further. It (0.23 g, 0.41 mmol) was dissolved in pyridine–water (9:1 v/v, 7 cm<sup>3</sup>) and zinc dust (0.2 g) was added to the vigorously stirred mixture. After 30 min the reaction mixture was diluted with EtOAc, filtered and treated with 1 mol dm<sup>-3</sup> aq. NaHCO<sub>3</sub> (EtOAc layer was discarded). The aqueous layer was acidified with cold, dil. HCl and then extracted with EtOAc. The ethyl acetate layer was washed with brine, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and the solvent was removed. The residue was taken up in MeOH–EtOAc and a slight excess of diethylamine was added; the precipitate was filtered off and washed successively with hot EtOAc and dry diethyl ether. Yield of **compound 11** was 0.095 g (42%), m.p. 180 °C and then 192–194 °C;  $\nu_{max}/cm^{-1}$  3413, 2985, 2703, 1732, 1698, 1441, 1379, 1227, 1198 and 1053;  $\delta_H$ (D<sub>2</sub>O) 0.96 (3 H, s, Me), 1.03 (3 H, s, Me), 1.26 [6 H, t,  $J$  7.2, N(CH<sub>2</sub>Me)<sub>2</sub>], 1.4 (3 H, s, Me), 1.43 (3 H, s, Me), 1.64 (3 H, s, Me), 2.17 (3 H, s, Ac), 2.27 (1 H, d,  $J$  2.7, 5-H), 2.58 (1 H, d,  $J_{gem}$  18, 12 $\beta$ -H), 3.04 [4 H, quartet,  $J$  7.2, N(CH<sub>2</sub>Me)<sub>2</sub>], 3.38 (1 H, d,  $J_{gem}$  18, 12 $\alpha$ -H), 4.37–4.57 (2 H, m, 1- and 7-H), 5.0 (1 H, dd,  $J_{cis}$  10.8,  $J_{gem}$  1.8, 15-H, Z), 5.23 (1 H, dd,  $J_{trans}$  18,  $J_{gem}$  1.8, 15-H, E), 5.83 (1 H, t,  $J$  3.6, 6-H) and 6.13 (1 H, dd,  $J_{trans}$  18,  $J_{cis}$  10.8, 14-H) (Found: C, 56.9; H, 8.4; N, 2.7. C<sub>26</sub>H<sub>46</sub>NO<sub>9</sub>P requires C, 57.02; H, 8.47; N, 2.56%).

7 $\beta$ -(tert-Butyldimethylsiloxy)-8,13-epoxy-6 $\beta$ ,9 $\alpha$ -dihydroxy-11-oxolabd-14-en-1 $\alpha$ -yl Diphenylphosphate 14.—8,13-Epoxy-1 $\alpha$ ,6 $\beta$ ,7 $\beta$ ,9 $\alpha$ -tetrahydroxylabd-14-en-11-one 13 (3.68 g, 10 mmol) and imidazole (1.7 g, 25 mmol) were dissolved in DMF (80 cm<sup>3</sup>) and to this was added TBDMSCl (1.96 g, 13 mmol). The mixture was heated at 35 °C for 2 h. The solvent was removed under reduced pressure and the residue was purified by flash chromatography with 10% EtOAc–C<sub>6</sub>H<sub>6</sub> to give an *intermediate* (3 g, 64%), m.p. 166–168 °C;  $\nu_{max}/cm^{-1}$  3003, 1702, 1282, 1219, 1117, 1041 and 923;  $\delta_H$  0.13 (3 H, s, SiMe), 0.17 (3 H, s, SiMe), 0.91 (9 H, s, Bu<sup>t</sup>), 1.05 (3 H, s, Me), 1.26 (3 H, s, Me), 1.38 (3 H, s, Me), 1.39 (3 H, s, Me), 1.56 (3 H, s, Me), 2.46 (1 H, d,  $J_{gem}$  17, 12 $\beta$ -H), 3.16 (1 H, s,  $J_{gem}$  17, 12 $\alpha$ -H), 4.04 (1 H, d,  $J$  4.53, 7-H), 4.2 (1 H, dd,  $J$  2.58 and 4.53, 6-H), 4.56 (1 H, br, 1-H), 4.95 (1 H, dd,  $J_{cis}$  11.1,  $J_{gem}$  1.54, 15-H, Z), 5.15 (1 H, dd,  $J_{trans}$  17.48,  $J_{gem}$  1.54, 15-H, E) and 6.05 (1 H, dd,  $J_{trans}$  17.48,  $J_{cis}$  11.1, 14-H) (Found: C, 64.9; H, 9.2. C<sub>25</sub>H<sub>46</sub>O<sub>6</sub>Si requires C, 64.73; H, 9.54%).

This compound (0.484 g, 1 mmol) was added to NaH (0.096 g, 4 mmol) suspended in THF (20 cm<sup>3</sup>) and the mixture was stirred for 15 min at room temperature. To this mixture at 0 °C was added diphenyl phosphorochloridate 0.268 g, 1 mmol) dropwise. The reaction mixture was brought to room temperature and was stirred overnight. The solvent was removed and the residue was taken up in EtOAc, washed successively with cold, dil. HCl and water and dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. After removal of the solvent under reduced pressure, the residue was purified by flash chromatography with 4% EtOAc–C<sub>6</sub>H<sub>6</sub> to give **compound 14** (0.348 g, 49%) as a

semisolid,  $\nu_{\max}/\text{cm}^{-1}$  3636, 3003, 1724, 1597, 1493, 1316, 1192 and 1164;  $\delta_{\text{H}}$  0.16 (3 H, s, SiMe), 0.17 (3 H, s, SiMe), 0.93 (9 H, s, Bu'), 1.09 (3 H, s, Me), 1.31 (3 H, s, Me), 1.41 (3 H, s, Me), 1.56 (3 H, s, Me), 1.61 (3 H, s, Me), 1.97 (1 H, d,  $J$  2.7, 5-H), 2.65 (1 H, d,  $J_{\text{gem}}$  18, 12 $\beta$ -H), 3.26 (1 H, d,  $J_{\text{gem}}$  18, 12 $\alpha$ -H), 4.11 (1 H, d,  $J$  3.6, 7-H), 4.27 (1 H, t,  $J$  2.7, 6-H), 5.01 (1 H, dd,  $J_{\text{cis}}$  10.8,  $J_{\text{gem}}$  1.8, 15-H, Z), 5.01 (1 H, br, 1-H), 5.24 (1 H, dd,  $J_{\text{trans}}$  17.1,  $J_{\text{gem}}$  1.8, 15-H, E), 6.01 (1 H, dd,  $J_{\text{trans}}$  17.1,  $J_{\text{cis}}$  10.8, 14-H) and 7.19–7.36 (10 H, m, Ph) (Found: C, 63.8; H, 7.7; P, 3.9.  $\text{C}_{38}\text{H}_{55}\text{O}_9\text{PSi}$  requires C, 63.84; H, 7.76; P, 4.33%).

8,13-Epoxy-1 $\alpha$ ,6 $\beta$ ,9 $\alpha$ -trihydroxy-11-oxolabd-14-en-7 $\beta$ -yl Bis-(2,2,2-trichloroethyl)phosphate **15**.—A solution of compound **13** (0.736 g, 2 mmol) in pyridine (15 cm<sup>3</sup>) was chilled to  $-10^\circ\text{C}$ . Bis-(2,2,2-trichloroethyl) phosphorochloridate (0.91 g, 2.4 mmol) was added to the above vigorously stirred mixture, which was stirred for a further 4 h at  $-10^\circ\text{C}$  before being taken up in EtOAc and washed successively with cold, 1 mol dm<sup>-3</sup> HCl and water. The EtOAc layer was dried over anhydrous  $\text{Na}_2\text{SO}_4$  and the solvent was removed. The oily residue was purified by flash chromatography with 20% EtOAc–light petroleum to give compound **15** (0.67 g, 47%), m.p. 155–157 $^\circ\text{C}$  (from EtOAc–light petroleum);  $\nu_{\max}/\text{cm}^{-1}$  3448, 3200, 1709, 1266, 1170 and 1096;  $\delta_{\text{H}}$  1.04 (3 H, s, Me), 1.26 (3 H, s, Me), 1.41 (6 H, s, 2  $\times$  Me), 1.7 (3 H, s, Me), 2.14 (1 H, d,  $J$  2.88, 5-H), 2.46 (1 H, d,  $J_{\text{gem}}$  17.1, 12 $\beta$ -H), 3.24 (1 H, d,  $J_{\text{gem}}$  17.1, 12 $\alpha$ -H), 4.64–4.71 (6 H, two broad peaks, 2  $\times$   $\text{CH}_2\text{CCl}_3$ , 1- and 6-H), 4.97 (1 H, dd,  $J_{\text{cis}}$  10.8,  $J_{\text{gem}}$  1.7, 15-H, Z), 5.03 (1 H, br, 7-H), 5.24 (1 H, dd,  $J_{\text{trans}}$  18,  $J_{\text{gem}}$  1.7, 15-H, E) and 6.14 (1 H, dd,  $J_{\text{trans}}$  18,  $J_{\text{cis}}$  10.8, 14-H) (Found: C, 40.6; H, 5.35; P, 3.95; Cl, 29.1.  $\text{C}_{24}\text{H}_{35}\text{Cl}_6\text{O}_9\text{P}$  requires C, 40.53; H, 4.96; P, 4.36; Cl, 29.91%).

8,13-Epoxy-1 $\alpha$ ,9 $\alpha$ -dihydroxy-11-oxolabd-14-ene-6 $\beta$ ,7 $\beta$ -diyl 2-Hydroxyethylcyclic Phosphate **16**.—To a vigorously stirred, chilled ( $-5$  to  $0^\circ\text{C}$ ) solution of compound **13** (3.68 g, 10 mmol) in pyridine (50 cm<sup>3</sup>) was added 2-chloro-2-oxo-1,3,2-dioxaphospholane (2.86 cm<sup>3</sup>, 20 mmol) dropwise. After 1 h at  $0^\circ\text{C}$  the mixture was kept in a refrigerator ( $5^\circ\text{C}$ ) for 12 h. It was then taken up in EtOAc (150 cm<sup>3</sup>) and washed successively with cold, 1 mol dm<sup>-3</sup> HCl and water. It was dried over anhydrous  $\text{Na}_2\text{SO}_4$ , the solvent was removed and the residue was purified by flash chromatography with 5% MeCN in  $\text{CHCl}_3$  to give compound **16** (0.71 g, 15%), m.p. 194–195 $^\circ\text{C}$  (EtOAc–light petroleum);  $\nu_{\max}/\text{cm}^{-1}$  3077, 2941, 1695, 1449, 1379, 1266, 1190, 1146 and 1086;  $\delta_{\text{H}}$  1.10 (3 H, s, Me), 1.19 (3 H, s, Me), 1.35 (3 H, s, Me), 1.41 (3 H, s, Me), 1.65 (3 H, s, Me), 2.34 (1 H, d,  $J_{\text{gem}}$  16.2, 12 $\beta$ -H), 2.44 (1 H, br, 5-H), 3.37 (1 H, d,  $J_{\text{gem}}$  16.2, 12 $\alpha$ -H), 3.70 (2 H, t,  $J$  5.4,  $\text{CH}_2\text{CH}_2\text{OH}$ ), 3.36 (2 H, quintet,  $J$  10.8 and 5.4,  $\text{CH}_2\text{CH}_2\text{OH}$ ), 4.57 (1 H, br, 1-H) 4.8 (1 H, d,  $J$  5.7, 7-H), 4.97 (1 H, dd,  $J_{\text{cis}}$  10.8,  $J_{\text{gem}}$  1.7, 15-H, Z), 5.07–5.14 (1 H, br, 6-H), 5.16 (1 H, dd,  $J_{\text{trans}}$  17.1,  $J_{\text{gem}}$  1.7, 15-H, E) and 6.4 (1 H, dd,  $J_{\text{trans}}$  17.1,  $J_{\text{cis}}$  10.8, 14-H) (Found: C, 54.1; H, 7.7; P, 6.4.  $\text{C}_{22}\text{H}_{35}\text{O}_9\text{P}\cdot\text{H}_2\text{O}$  requires C, 53.65; H, 7.57; P, 6.29%).

7 $\beta$ -Acetoxy-8,13-epoxy-6 $\beta$ ,9 $\alpha$ -dihydroxy-11-oxolabd-14-en-1 $\alpha$ -yl Dihydrogen Phosphate **17**.—Forskolin **1** (5.5 g, 13.4 mmol) was dissolved in a mixture of  $\text{CH}_2\text{Cl}_2$  (150 cm<sup>3</sup>) and ethyldiisopropylamine (10 cm<sup>3</sup>). To this vigorously stirred solution was added 1,2-dibromo-2-phenylethylphosphonic acid (4.0 g, 11.6 mmol). The clear solution was kept at room temperature for 48 h. The solvent was then removed under reduced pressure and the residue was partitioned between EtOAc and 1 mol dm<sup>-3</sup> aq.  $\text{NaHCO}_3$  (20 cm<sup>3</sup>). The aqueous layer was washed with EtOAc (2  $\times$  5 cm<sup>3</sup>), and the combined EtOAc portion was discarded. The aqueous layer was acidified with cold, 1 mol dm<sup>-3</sup> HCl to pH 2 and was then saturated with NaCl and extracted with EtOAc (5  $\times$  15 cm<sup>3</sup>). The EtOAc layer was washed with brine, then dried over anhydrous

$\text{Na}_2\text{SO}_4$  and the solvent was removed. The residue was crystallized from EtOAc–light petroleum to give compound **17** (2 g, 30%), m.p. 149–151 $^\circ\text{C}$ ;  $\nu_{\max}/\text{cm}^{-1}$  3571, 2958, 1724, 1369, 1257 and 1113;  $\delta_{\text{H}}$ ( $\text{CDCl}_3 + \text{CD}_3\text{OD}$ ) 1.04 (3 H, s, Me), 1.24 (3 H, s, Me), 1.33 (3 H, s, Me), 1.53 (3 H, s, Me), 1.66 (3 H, s, Me), 2.15 (3 H, s, Ac), 2.2 (1 H, d,  $J$  3.6, 5-H), 2.27 (1 H, d,  $J_{\text{gem}}$  15.12, 12 $\beta$ -H), 3.26 (1 H, d,  $J_{\text{gem}}$  15.12, 12 $\alpha$ -H), 4.46 (1 H, t,  $J$  3.6, 6-H), 4.9 (1 H, dd,  $J_{\text{cis}}$  10.8,  $J_{\text{gem}}$  1.8, 15-H, Z), 5.04 (1 H, br s, 1-H), 5.19 (1 H, dd,  $J_{\text{trans}}$  18,  $J_{\text{gem}}$  1.8, 15-H, E), 5.33 (1 H, d,  $J$  3.6, 7-H) and 5.94 (1 H, dd,  $J_{\text{trans}}$  18,  $J_{\text{cis}}$  10.8, 14-H) (Found: C, 53.5; H, 7.4; P, 6.4.  $\text{C}_{22}\text{H}_{35}\text{O}_{10}\text{P}$  requires C, 53.87; H, 7.19; P, 6.32%).

8,13-Epoxy-1 $\alpha$ ,6 $\beta$ ,9 $\alpha$ -trihydroxy-11-oxolabd-14-en-7 $\beta$ -yl Di-hydrogen Phosphate **18**.—To a solution of compound **13** (3.68 g, 10 mmol) in a mixture of  $\text{CH}_2\text{Cl}_2$  (50 cm<sup>3</sup>) and ethyldiisopropylamine (5 cm<sup>3</sup>) was added 1,2-dibromo-2-phenylethylphosphonic acid (3.44 g, 10 mmol). The clear solution was kept at room temperature for 48 h. The solvent was removed under reduced pressure and the solid residue was partitioned between EtOAc and water. The aqueous layer was extracted with EtOAc (2  $\times$  10 cm<sup>3</sup>), the EtOAc portion was discarded and the aqueous layer was acidified with cold, 1 mol dm<sup>-3</sup> HCl to pH 2, and then extracted with EtOAc (3  $\times$  15 cm<sup>3</sup>). The EtOAc layer was washed with brine, then dried over anhydrous  $\text{Na}_2\text{SO}_4$  and the solvent was removed. The solid obtained was crystallized from EtOAc–light petroleum to give compound **18** (1.3 g, 29%), m.p. 149 $^\circ\text{C}$  (decomp.), 185–189 $^\circ\text{C}$ ;  $\nu_{\max}/\text{cm}^{-1}$  3371, 2941, 1709, 1449, 1379, 1250 and 1169;  $\delta_{\text{H}}$ ( $\text{CDCl}_3 + \text{CD}_3\text{OD}$ ) 1.06 (3 H, s, Me), 1.29 (3 H, s, Me), 1.44 (3 H, s, Me), 1.46 (3 H, s, Me), 1.76 (3 H, s, Me), 2.11 (1 H, d,  $J$  2.88, 5-H), 2.47 (1 H, d,  $J_{\text{gem}}$  17.1, 12 $\beta$ -H), 3.29 (1 H, d,  $J_{\text{gem}}$  17.1, 12 $\alpha$ -H), 4.51–4.79 (3 H, m, 1-, 6- and 7-H), 5.01 (1 H, dd,  $J_{\text{cis}}$  10.8,  $J_{\text{gem}}$  1.7, 15-H, Z), 5.23 (1 H, dd,  $J_{\text{trans}}$  18,  $J_{\text{gem}}$  1.7, 15-H, E) and 6.19 (1 H, dd,  $J_{\text{trans}}$  18,  $J_{\text{cis}}$  10.8, 14-H) (Found: C, 53.1; H, 7.6; P, 7.0.  $\text{C}_{20}\text{H}_{33}\text{O}_9\text{P}$  requires C, 53.56; H, 7.42; P, 6.91%).

7 $\beta$ -Acetoxy-8,13-epoxy-6 $\beta$ -hydroxy-11-oxolabd-14-ene-1 $\alpha$ ,9 $\alpha$ -diyl Cyclophosphate Diethylamine Salt **19**.—Compound **2** (0.47 g, 0.96 mmol) was dissolved in 1,4-dioxane (5 cm<sup>3</sup>)–water (5 cm<sup>3</sup>) and  $\text{Na}_2\text{HPO}_4$  (0.2 g) and  $\text{KH}_2\text{PO}_4$  (0.05 g) were added. The mixture was refluxed for 2 h on a steam-bath. The solvent was removed under reduced pressure. The residue was diluted with water and made alkaline by addition of saturated aq.  $\text{NaHCO}_3$ . The aqueous layer was extracted with diethyl ether (2  $\times$  15 cm<sup>3</sup>) and the clear aqueous layer was acidified with cold, 1 mol dm<sup>-3</sup> HCl. The separated oil was extracted with EtOAc (4  $\times$  5 cm<sup>3</sup>). The EtOAc layer was washed with brine, dried over anhydrous  $\text{Na}_2\text{SO}_4$ , and the solvent was removed, under reduced pressure, below 40 $^\circ\text{C}$ . The residue was dried in a vacuum desiccator over KOH overnight. It was then dissolved in diethyl ether and freshly distilled diethylamine was added. The precipitate thus formed was filtered off, washed with dry diethyl ether and recrystallized from MeOH–Et<sub>2</sub>O to give compound **19** (0.4 g, 76%), m.p. 228–230 $^\circ\text{C}$ ;  $\nu_{\max}/\text{cm}^{-1}$  3419, 2963, 2941, 1714, 1366, 1231, 1176, 1163 and 1099;  $\delta_{\text{H}}$  1.01 (3 H, s, Me), 1.29 (12 H, br, 2  $\times$  Me + 2  $\text{NCH}_2\text{Me}$ ), 1.55 (3 H, s, Me), 1.67 (3 H, s, Me), 2.10 (1 H merged with Ac, 5-H), 2.14 (3 H, s, Ac), 2.32 (1 H, d,  $J_{\text{gem}}$  17.1, 12 $\beta$ -H), 2.83 (4 H, quartet,  $J$  7.2,  $\text{MeCH}_2\text{N}$ ), 3.71 (1 H, d,  $J_{\text{gem}}$  17.1, 12 $\alpha$ -H), 4.43 (1 H, t,  $J$  2.88, 6-H), 4.54 (1 H, br, 1-H), 4.84 (1 H, dd,  $J_{\text{cis}}$  10.8,  $J_{\text{gem}}$  1.8, 15-H, Z), 5.19 (1 H, dd,  $J_{\text{trans}}$  18,  $J_{\text{gem}}$  1.8, 15-H, E), 5.4 (1 H, d,  $J$  3.6, 7-H), 5.91 (1 H, dd,  $J_{\text{trans}}$  18,  $J_{\text{cis}}$  10.8, 14-H) (Found: C, 56.9; H, 8.1; N, 2.7; P, 5.4.  $\text{C}_{24}\text{H}_{44}\text{NO}_9\text{P}$  requires C, 57.23; H, 8.13; N, 2.57; P, 5.68%).

7 $\beta$ -Acetoxy-8,13-epoxy-6 $\beta$ -hydroxy-11-oxolabd-14-ene-1 $\alpha$ ,9 $\alpha$ -diyl N-[3-(Dimethylamino)propyl]cyclophosphoramidate Hydrochloride **20**.—Compound **3** (0.6 g, 1.22 mmol) was dis-

solved in THF (6 cm<sup>3</sup>) and 3-(dimethylamino)propylamine (4 cm<sup>3</sup>) was added. The clear solution was kept at room temperature for 2 h. The reaction mixture was diluted with EtOAc and washed with water until neutral. The EtOAc layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. The solvent was removed under reduced pressure. The residual oil was purified by flash chromatography over silica gel with CHCl<sub>3</sub> and then 2–10% MeOH–CHCl<sub>3</sub> as eluent. The pure material was taken up in dry MeOH and HCl–diethyl ether was added to form the hydrochloride. The precipitate formed was filtered off, washed with dry diethyl ether and crystallized from acetone–diethyl ether to give **compound 20** (0.29 g, 40%), m.p. 217–218 °C;  $\nu_{\max}/\text{cm}^{-1}$  3419, 2941, 2685, 1717, 1262, 1242 and 1117;  $\delta_{\text{H}}(\text{CDCl}_3 + \text{CD}_3\text{OD})$  1.03 (3 H, s, Me), 1.29 (6 H, s, 2 × Me), 1.71 (3 H, s, Me), 1.86 (3 H, s, Me), 2.15 (3 H, s, Ac), 2.2–3.4 (14 H, m, 12-H<sub>2</sub> + CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>NMe<sub>2</sub>) 4.44 (1 H, t, J 3.24, 6-H), 4.61 (1 H, br, 1-H), 4.96 (1 H, dd,  $J_{\text{cis}}$  10.13,  $J_{\text{gem}}$  2, 15-H, Z), 5.09 (1 H, merged, 7-H), 5.19 (1 H, dd,  $J_{\text{trans}}$  17.2,  $J_{\text{gem}}$  2.0, 15-H, E) and 5.87 (1 H, dd,  $J_{\text{trans}}$  17.2,  $J_{\text{cis}}$  10.13, 14-H) (Found: C, 53.1; H, 8.0; N, 4.5; Cl, 5.6. C<sub>27</sub>H<sub>46</sub>ClN<sub>2</sub>O<sub>8</sub>P·H<sub>2</sub>O requires C, 53.06; H, 7.75; N, 4.59; Cl, 5.80%).

7 $\beta$ -Acetoxy-8,13-epoxy-6 $\beta$ -hydroxy-11-oxolabd-14-enc-1 $\alpha$ , 9 $\alpha$ -diyl N-(4-methylpiperazino)cyclophosphoramidate Hydrochloride **21**.—Compound **3** (1.0 g, 2.04 mmol) was dissolved in *N*-methylpiperazine (10 cm<sup>3</sup>) and the solution was kept at room temperature for 48 h before being diluted with EtOAc and washed with water until neutral. The solution was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and the solvent was removed under reduced pressure. The residue was purified by flash chromatography with 7.5% MeOH–CHCl<sub>3</sub> and the product was dissolved in EtOAc and HCl in diethyl ether was added. The precipitate thus formed was filtered off, and washed successively with dry EtOAc and diethyl ether to afford **compound 21** (0.46 g, 38%), m.p. 210–212 °C;  $\nu_{\max}/\text{cm}^{-1}$  3389, 2898, 1705, 1447, 1363, 1242, 1190, 1180, 1095 and 1053;  $\delta_{\text{H}}$  1.09 (3 H, s, Me), 1.24 (3 H, s, Me), 1.33 (3 H, s, Me), 1.51 (3 H, s, Me), 1.69 (3 H, s, Me), 2.13 (3 H, s, Ac), 2.25–3.16 (13 H, m, 2 × OH, 5-H, 12-H<sub>2</sub>, 4 × CH<sub>2</sub>), 4.51 (1 H, t, J 3.13, 6-H), 4.97 (1 H, dd,  $J_{\text{cis}}$  10.8,  $J_{\text{gem}}$  1.8, 15-H, Z), 5.03 (1 H, br, 1-H), 5.36 (1 H, dd,  $J_{\text{trans}}$  18,  $J_{\text{gem}}$  1.8,

15-H, E), 5.39 (1 H, d, J 3.6, 7-H), 5.83 (1 H, dd,  $J_{\text{trans}}$  18,  $J_{\text{cis}}$  10.8, 14-H) (Found: C, 52.0; H, 7.5; N, 4.2; Cl, 6.05. C<sub>27</sub>H<sub>44</sub>ClN<sub>2</sub>O<sub>8</sub>P·2H<sub>2</sub>O requires C, 51.71; H, 7.71; N, 4.47; Cl, 5.65%).

### Acknowledgements

We gratefully acknowledge Dr. R. H. Rupp for his interest and encouragement, the late Dr. A. N. Dohadwalla, the late Dr. N. K. Dadkar and their group for pharmacological screening, Dr. P. K. Inamdar and his group for analytical data, and Mrs. Amanda Nogueira for typing the manuscript.

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Paper 2/01212J

Received 5th March 1992

Accepted 27th March 1992